

Synthesis of Functionally Graded Materials by Electrophoretic Deposition and Sintering

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- **Committee members**

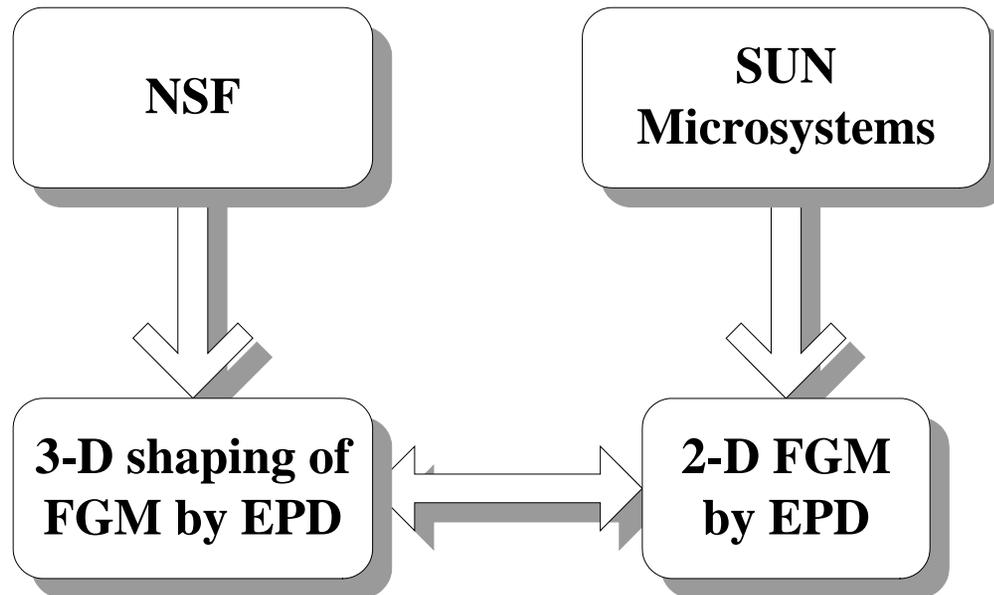
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Sponsors

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Fabrication routes for FGM

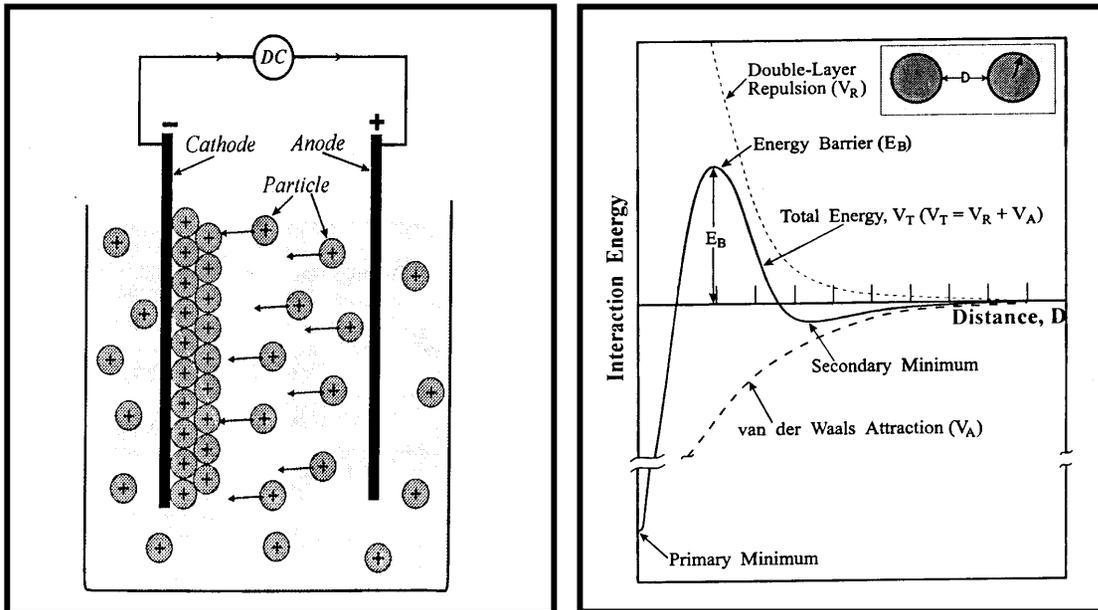
- **Constructive processes**
 - *Conventional solid state powder deposition*
 - *Liquid phase sintering*
 - *Infiltration*
 - *Reactive powder processes*
 - *Plasma spray forming*
 - *Laser cladding*
 - *Electroforming*
 - *Vapor deposition*
 - *Lamination processes*
- **Transport based processes**
 - *Mass transport processes*
 - *Thermal processes*
 - *Setting and centrifugal separation*

Electrophoretic Deposition (EPD)

EPD is a powder processing technology based on colloids

3 steps of Electrophoretic deposition

- Particle surface charging in solvent
- Particle migration under external electric field
- Particle coagulation at electrode



Picture by courtesy
of P. Sarkar and P.S.
Nicholson

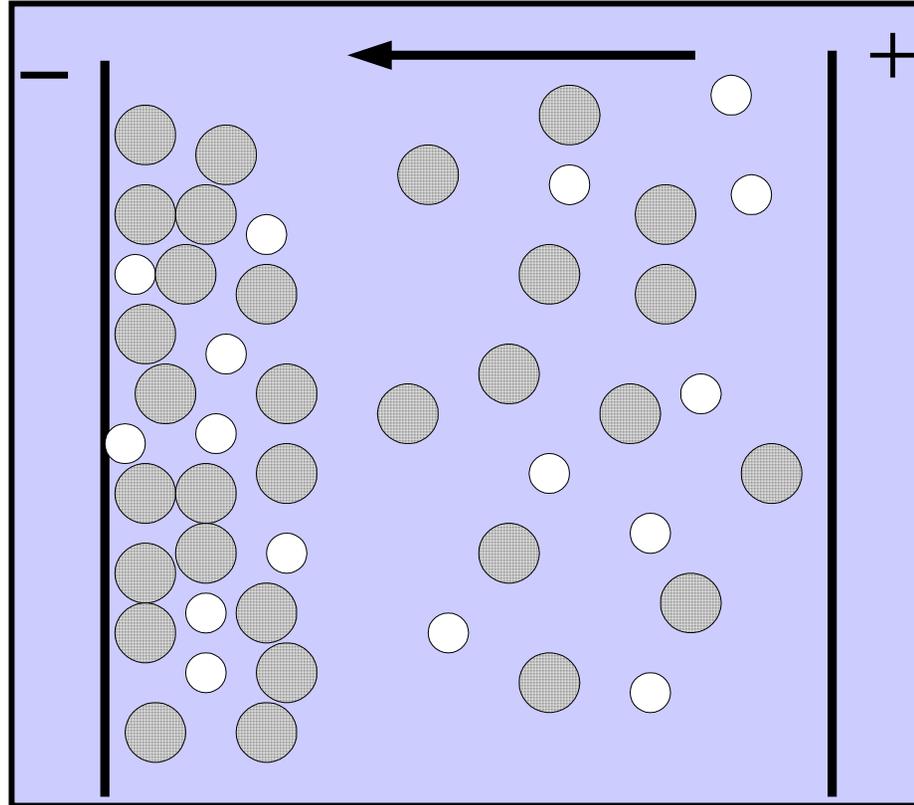
Advantages of EPD

- **Fast and convenient, easy to scale up**
- **Low investment and high flexibility (almost any materials: metals, ceramics and polymers)**
- **Capable of producing thick, thin films (ranging from micrometers to centimeters) and 3-D complex geometries**
- **Easy to produce composite materials with precisely tailored properties**

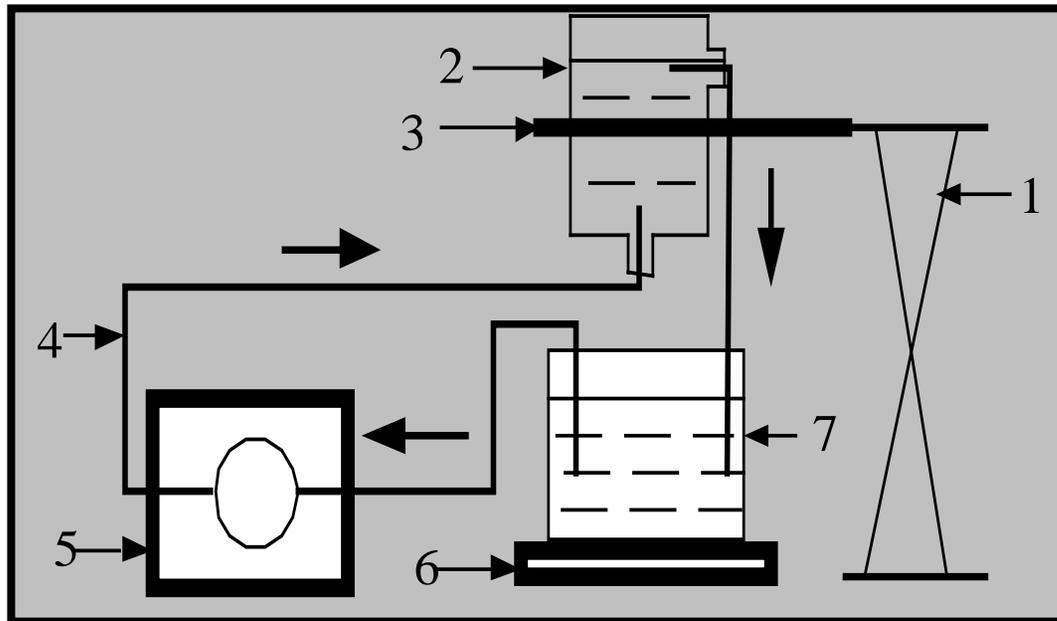
Research objectives

- **Net-shape manufacturing by sintering → inverse sintering problem solution: green specimen with special shape and with special (composite) structure;**
- **Fabrication of a special shape functionally structured (graded) green specimen → EPD;**
- **An ambient temperature processing technology for FGM fabrication as a parallel problem → the EPD-EP approach provides application in electronic packaging.**

Fabrication of FGM by EPD



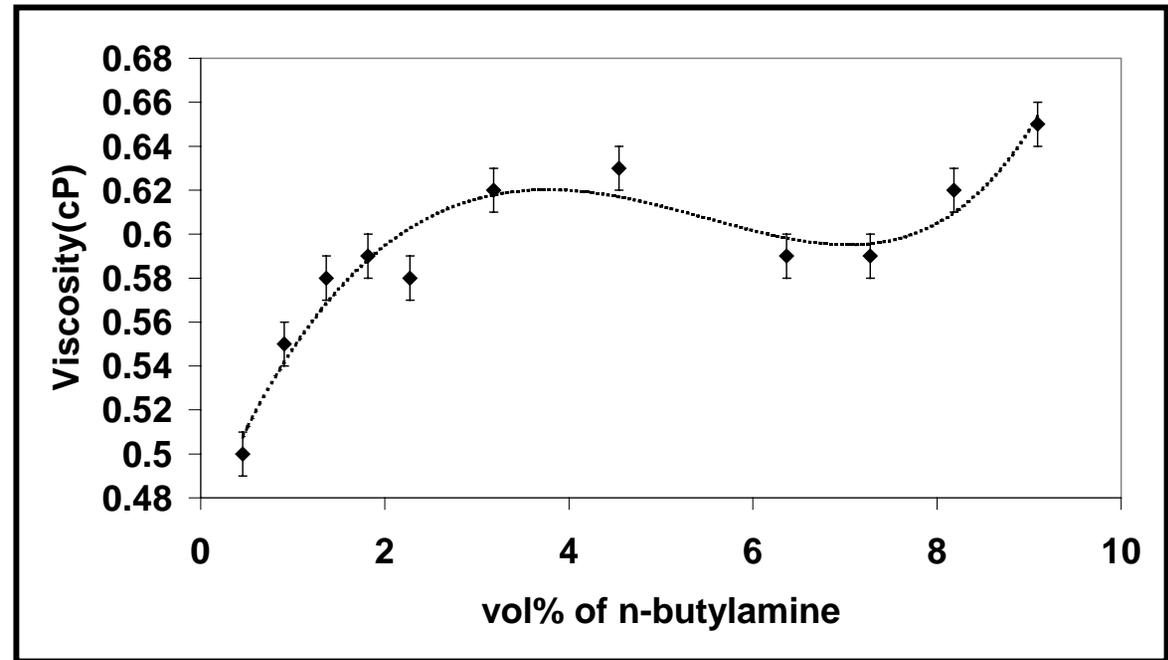
Equipment for EPD



1. Adjustable stand
2. Deposition cell
3. Deposition cell holder
4. Tubes
5. Pump
6. Magnetic stirrer
7. Suspension container

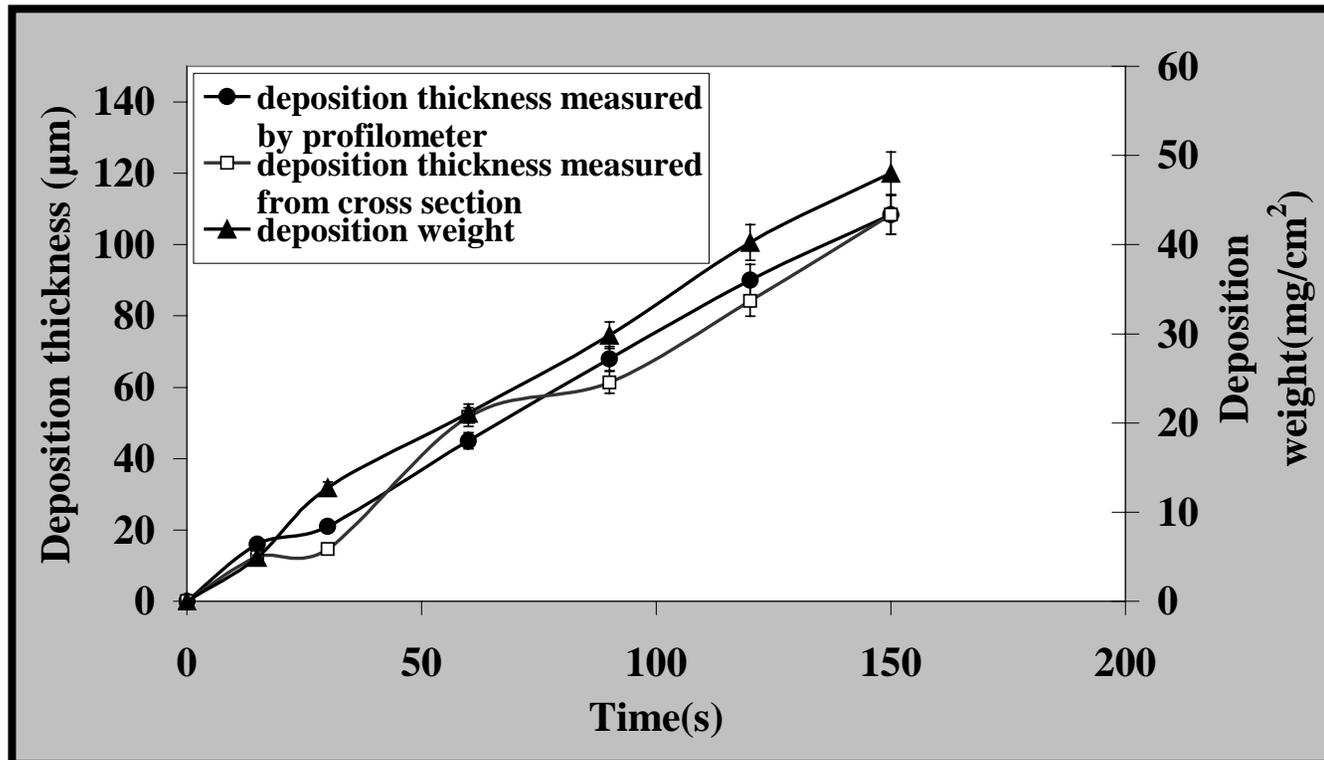
Characteristics of EPD suspension

- The suspension is acetone. N-butylamine was added to enhance particle charging
- The optimal concentration of n-butylamine was determined by viscosity measurement to make the suspension stable

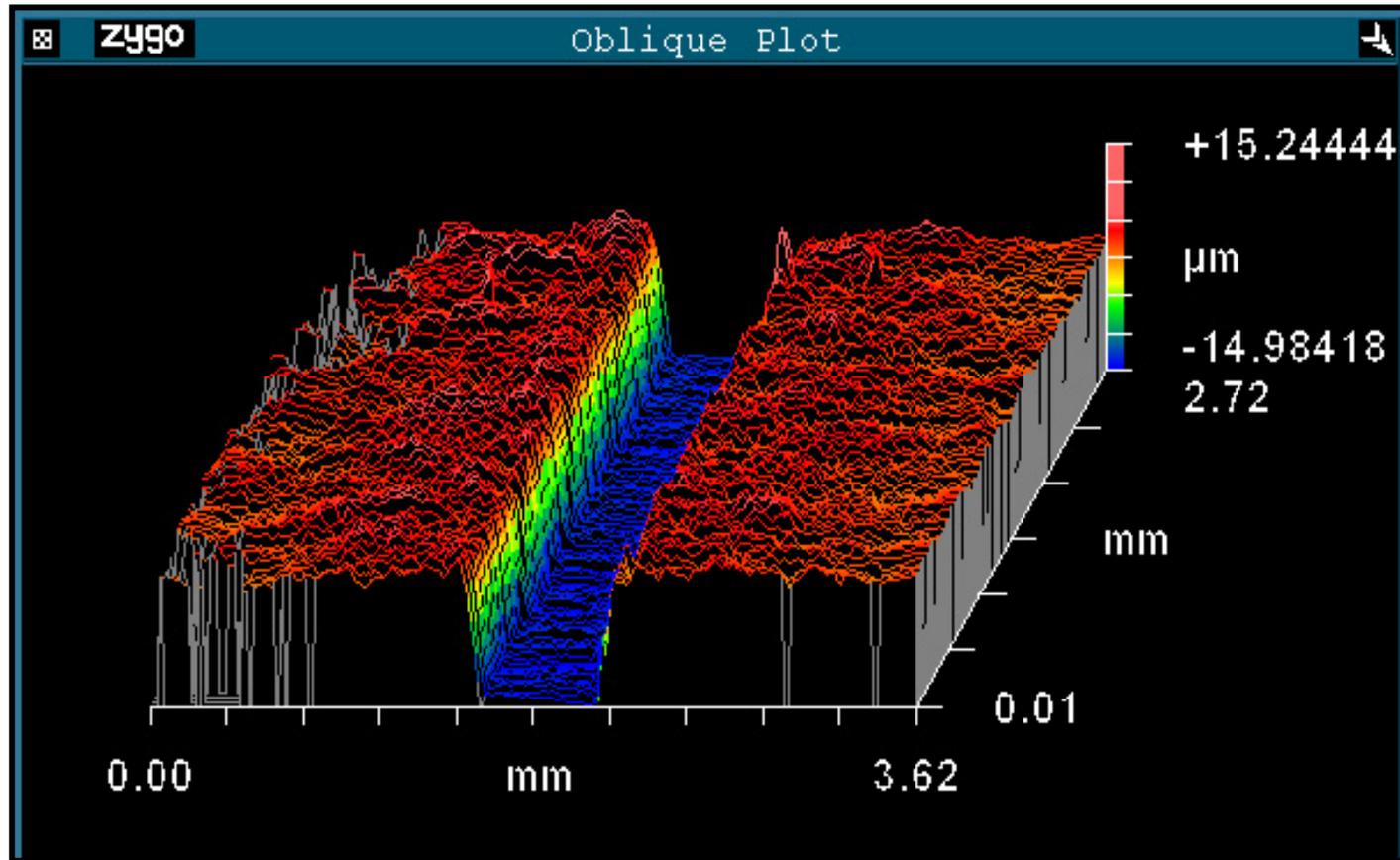


Kinetics of EPD (thin alumina film)

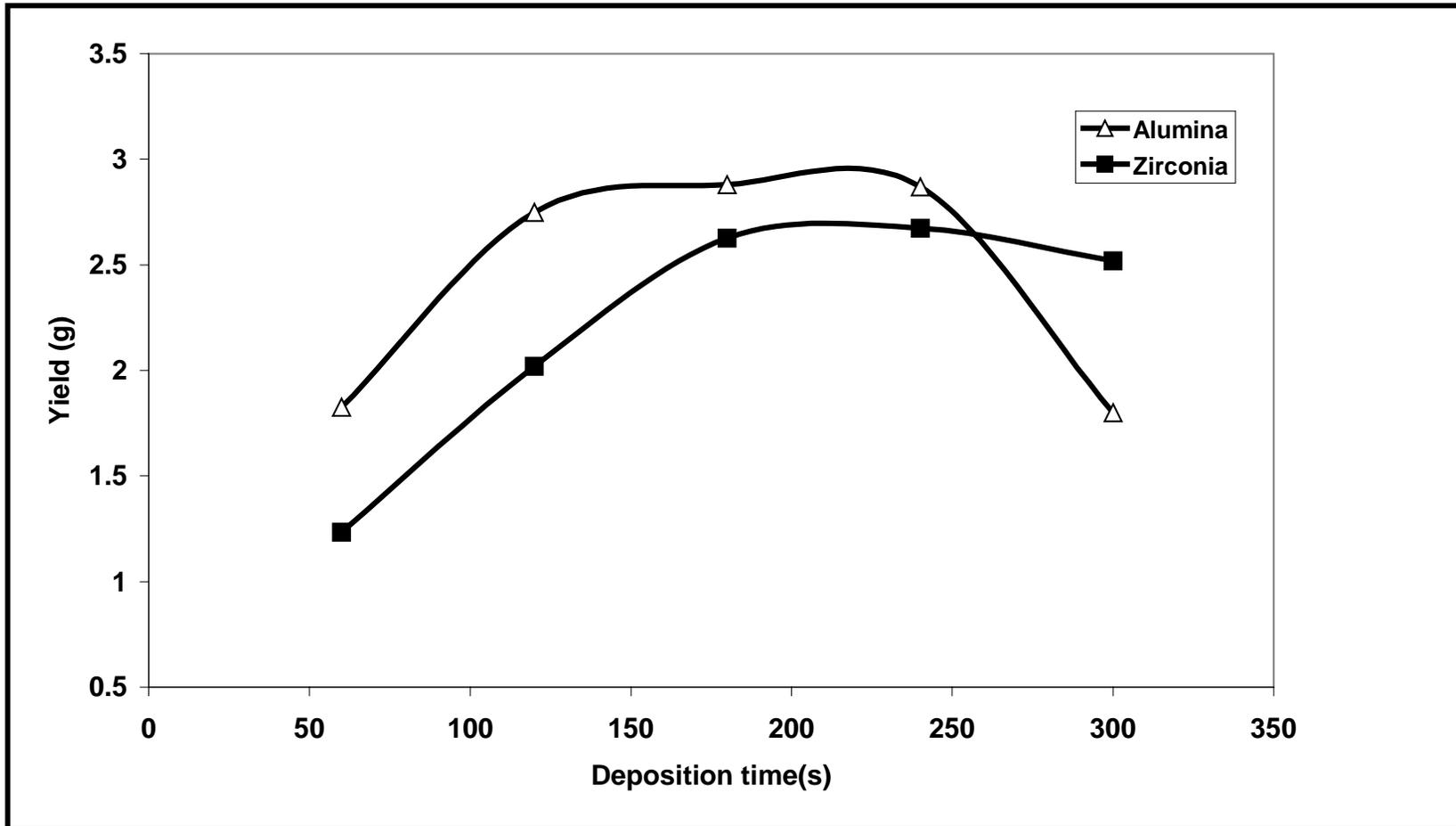
Objective: in order to control the thickness of EPD



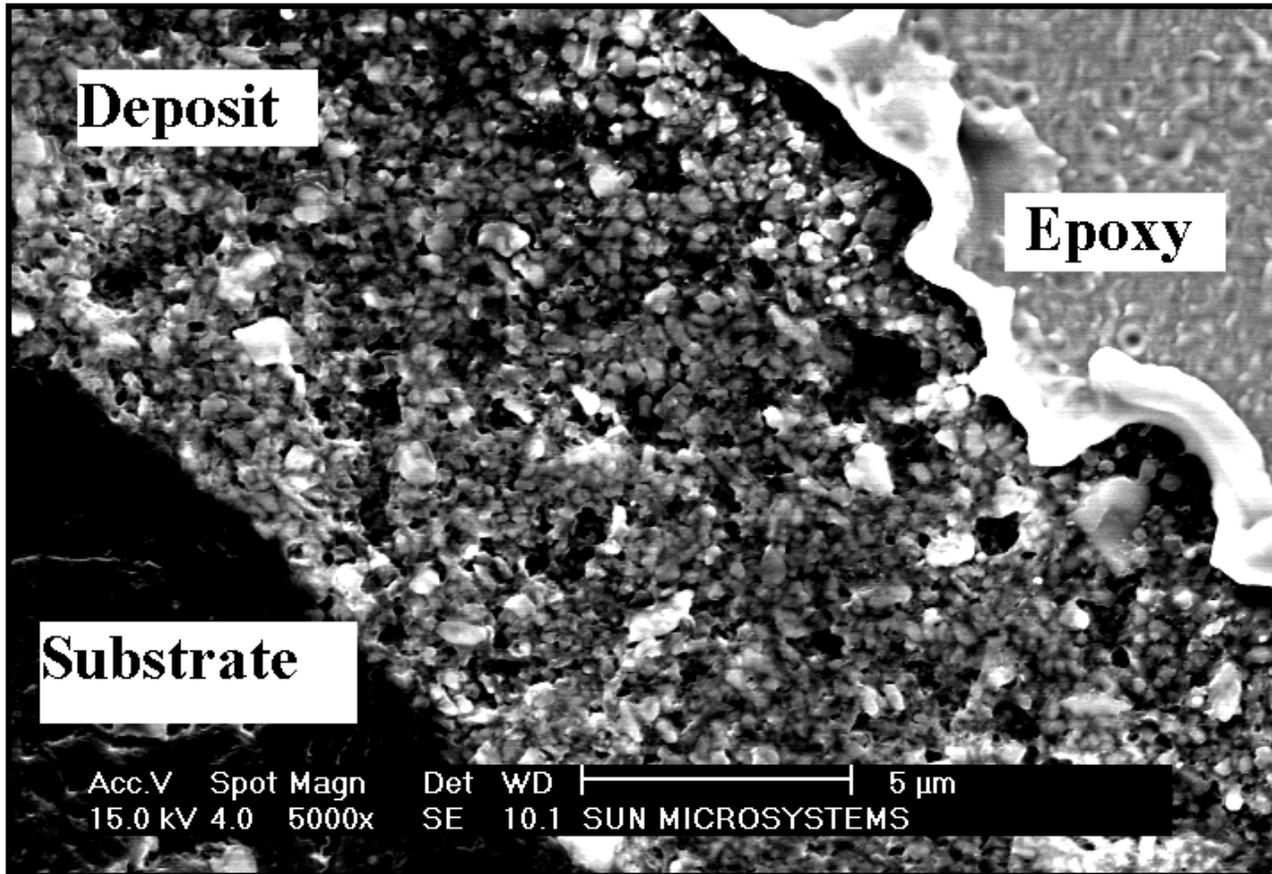
Measurement of EPD deposit thickness



Kinetics of EPD (thick alumina and zirconia deposits)



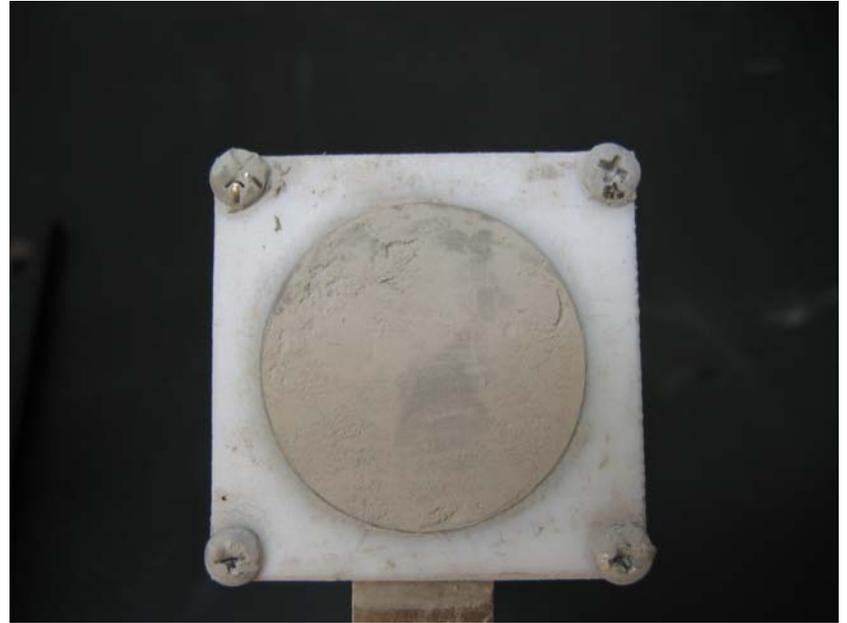
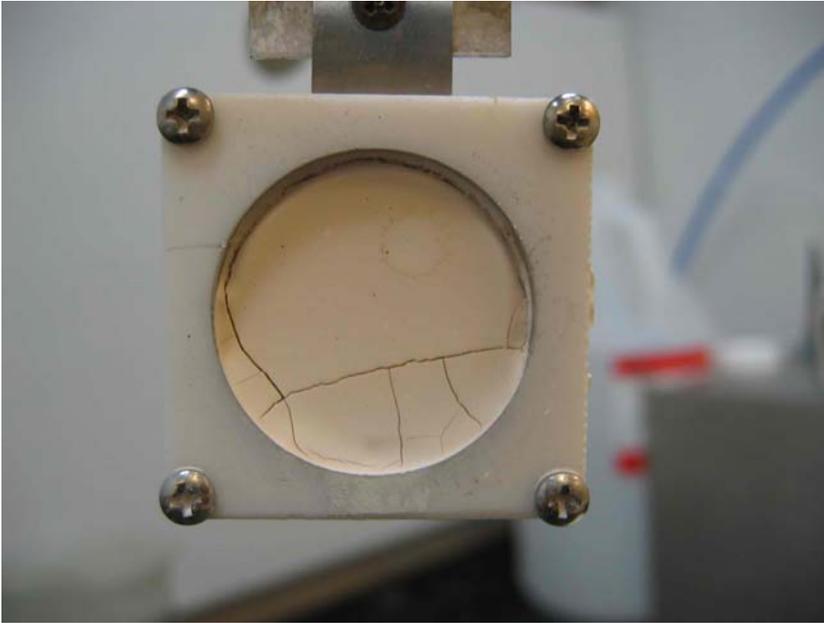
Cross section SEM of Alumina deposit obtained by EPD



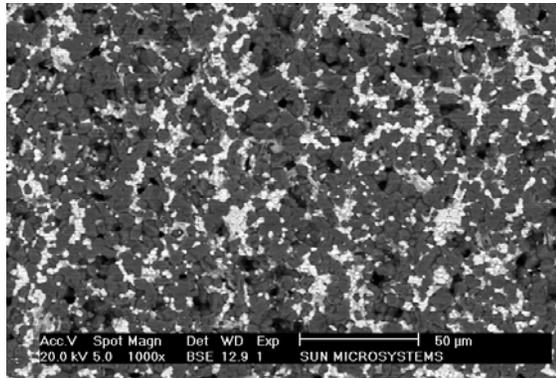
3-D EPD shaping



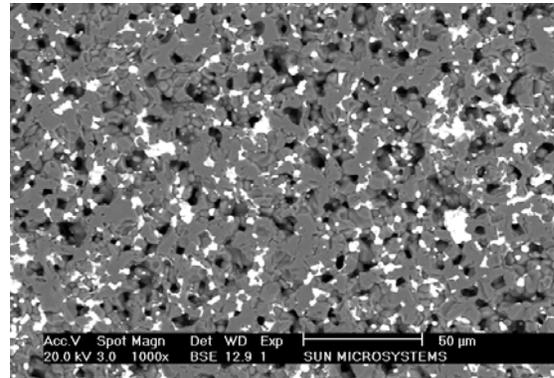
Cracking during drying



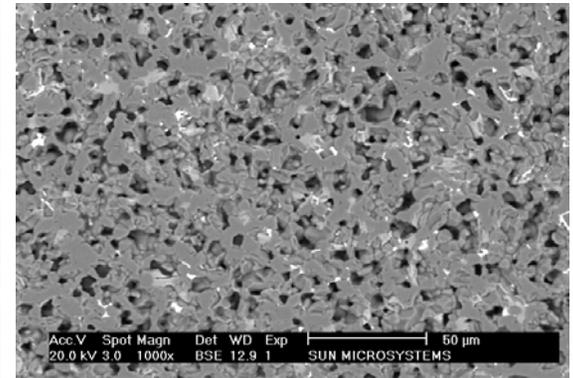
Microstructure of FGM cylinder



Zirconia rich side



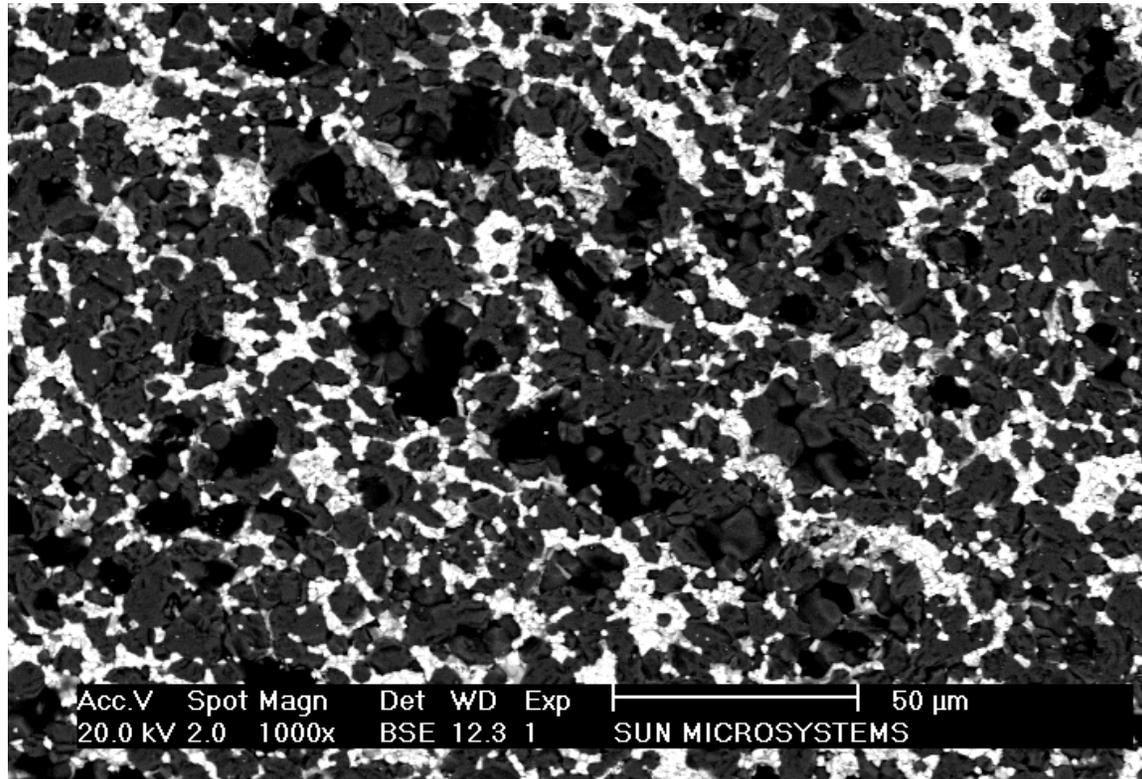
**Intermediate
layer**



**Alumina rich
side**

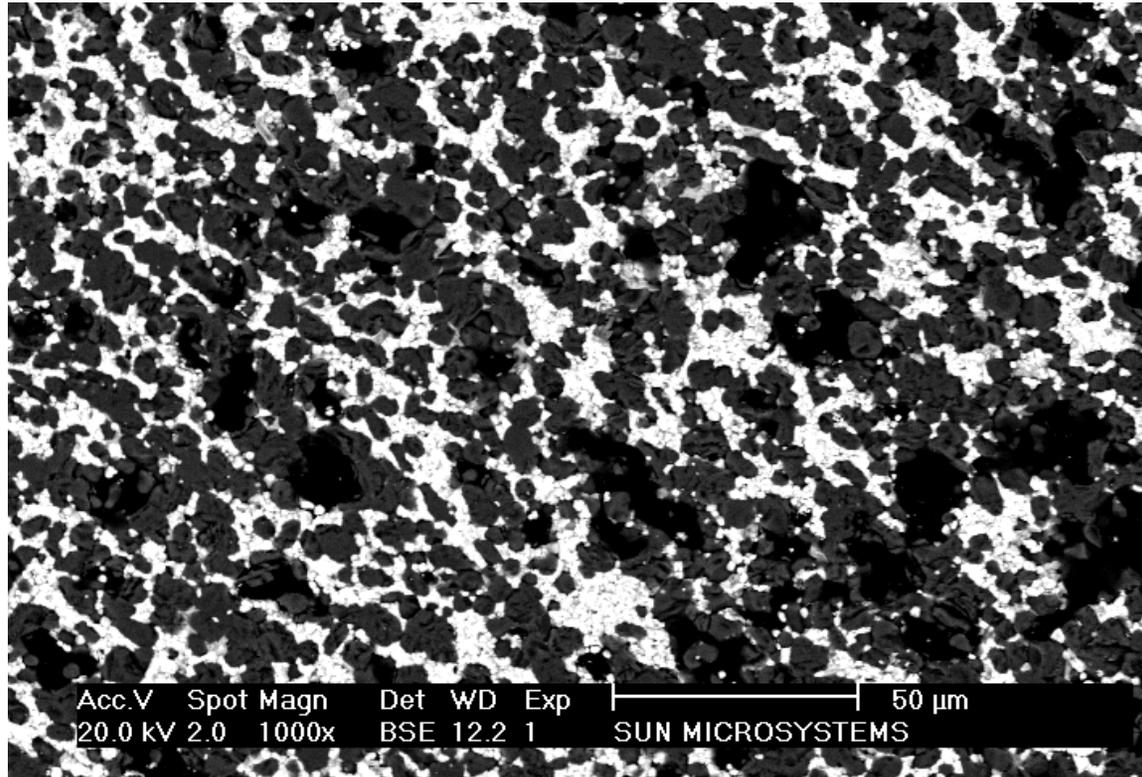
SEM pictures of different positions of FGM: white spots are zirconia, gray spots are alumina

Microstructure of FGM disks



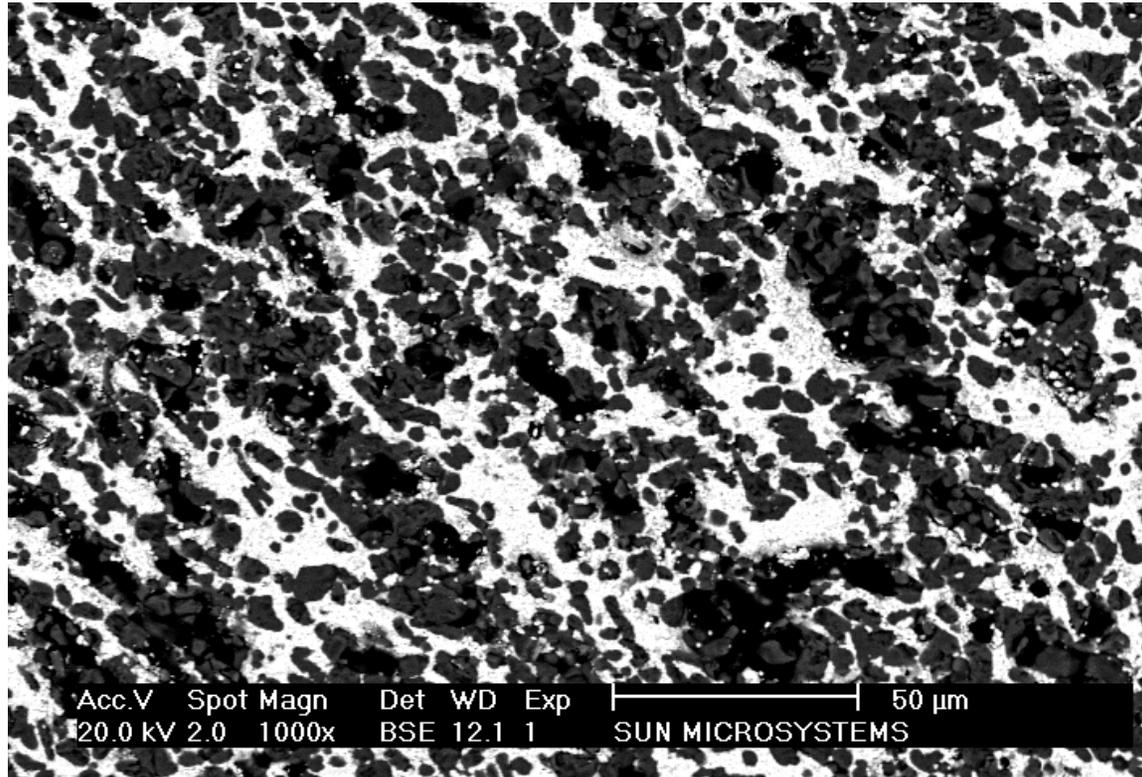
**Alumina rich side of an FGM disk
(FGM02)**

Microstructure of FGM disks



**Intermediate part of an FGM disk
(FGM02)**

Microstructure of FGM disks



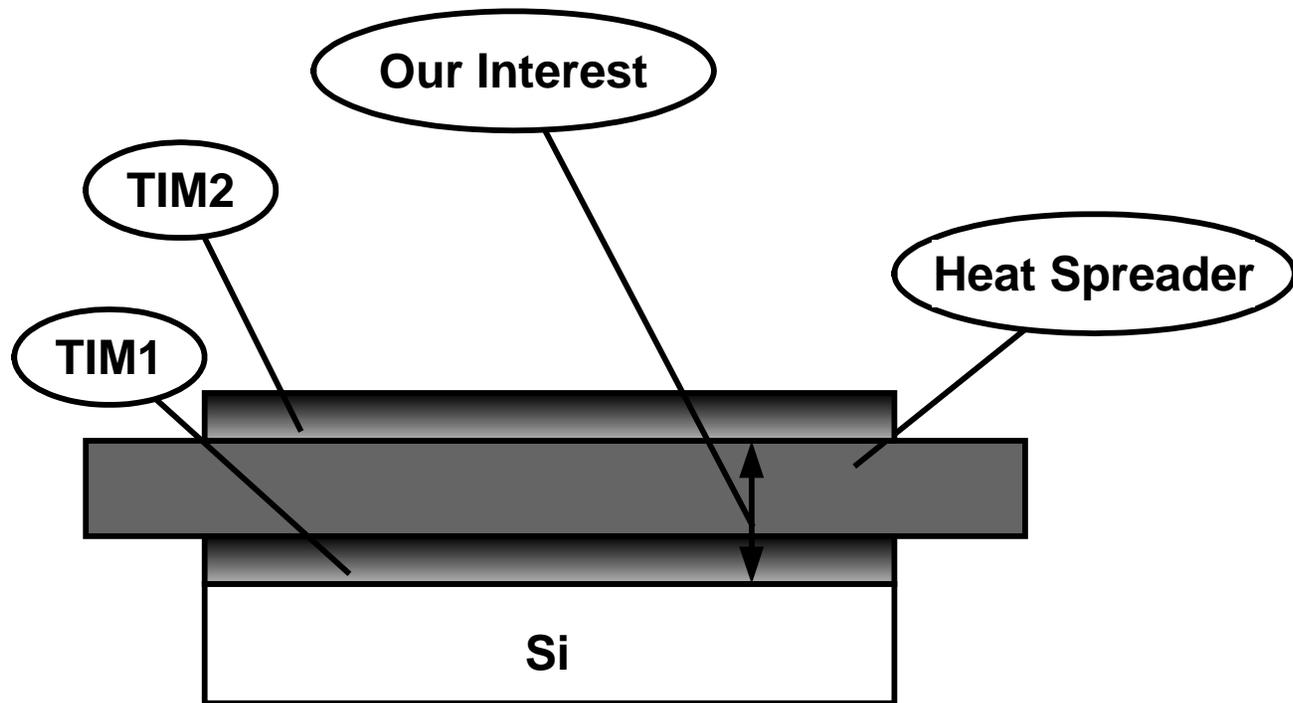
**Zirconia rich side of an FGM disk
(FGM02)**

Shape distortion of FGM after sintering



Background

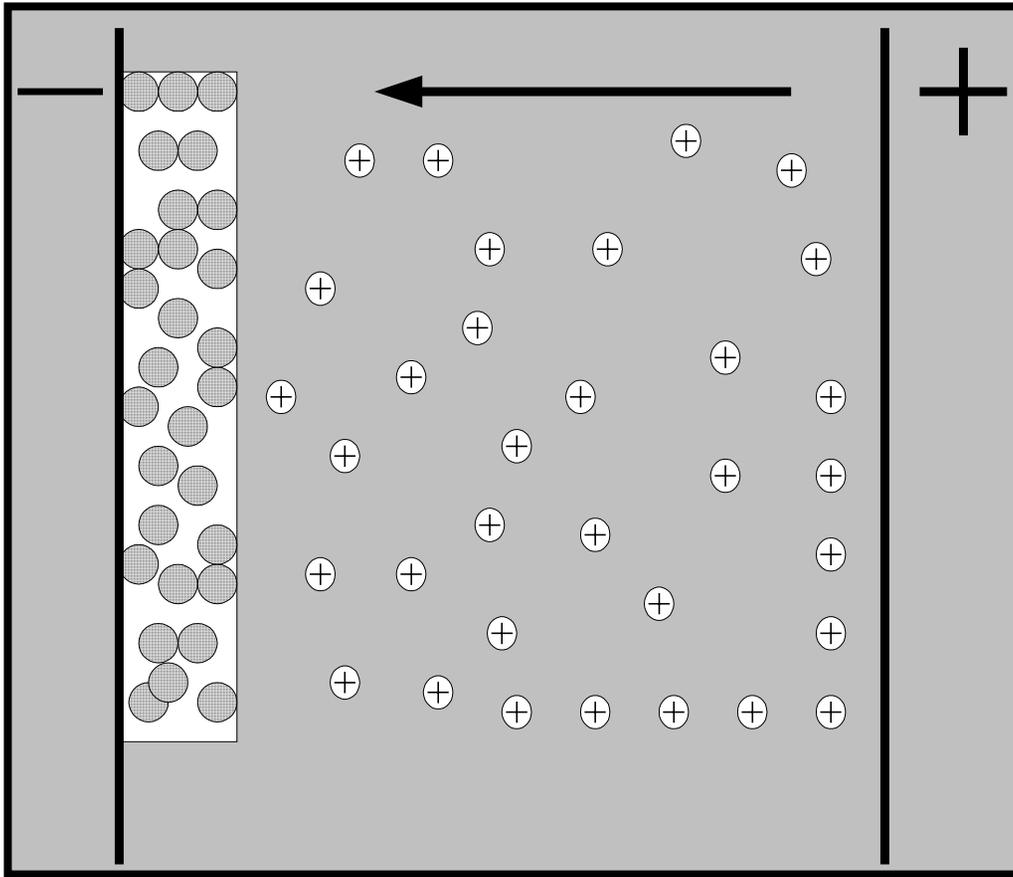
- High thermal conductivity
- Resolve CTE mismatch



Requirements for fabrication of TIM

- **Thermal properties of TIM should fall between silicon and copper: composite of metal and ceramics should be used**
- **Synthesis method should avoid high temperature processing**
- **Our approach: Electrophoretic deposition (deposition of ceramic particles) and electroplating (deposition of metal) to produce TIM without thermal processing**

The approach-sequential deposition



**Sequential Deposition:
EPD+Electroplating
in two steps**

The Main Constitutive Relationship

$$\sigma_{ij} = \frac{\sigma(W)}{W} \left[\varphi \dot{e}_{ij} + \left(\psi - \frac{1}{3} \varphi \right) \dot{e} \delta_{ij} \right] + P_L \delta_{ij}$$

Strain rate component \dot{e}_{ij}
 Bulk modulus: Resistance to the volume change function of porosity
 Shear modulus: Resistance to the shape change function of porosity
 Generalized viscosity: corresponds to the constitutive properties of particle material
 Volume strain rate \dot{e}
 Effective sintering stress: function of porosity

externally applied stresses material resistance sintering stresses

Linear Viscous: $\sigma(W) = 2\eta_0 W$

Rigid Plastic: $\sigma(W) = \sigma_y$

Power Law Creep: $\sigma(W) = AW^m$

$$W = \frac{1}{\sqrt{1-\theta}} \sqrt{\varphi \dot{\gamma} + \psi \dot{e}^2}$$

Formulations to model sintering of composite and FGM

Linear viscous case:
$$\sigma_{ij} = 2\eta_0 \left[\varphi \dot{\epsilon}_{ij} + \left(\psi - \frac{1}{3} \varphi \right) \dot{\epsilon} \delta_{ij} \right] + P_L$$

Skorohod model:
$$P_L = \frac{3\alpha}{r} (1 - \theta)^2$$

Sintering stress for composite materials:

$$P_L = -\frac{\alpha\gamma(1-\theta)N_c}{4} \left\{ \begin{array}{l} \frac{\phi_s c_{ls}}{R_s(\phi_l + c_{ls}\phi_s)} \left[\phi_s + \frac{(1 - \frac{\sqrt{3}}{2})\phi_l(1 + c_{ls})}{1 + c_{ls} - \sqrt{1 + 2c_{ls}}} \right] + \\ \frac{\phi_l c_{sl}}{R_l(\phi_s + c_{sl}\phi_l)} \left[\phi_l + \frac{(1 - \frac{\sqrt{3}}{2})\phi_s(1 + c_{sl})}{1 + c_{sl} - \sqrt{1 + 2c_{sl}}} \right] \end{array} \right\}$$

Formulations to model sintering of composite and FGM (cont'd)

Densification rate of mixed alumina and zirconia (from experimental results of Raj*):

$$\dot{\rho} = A \frac{\exp(-Q/R_g T)}{T} \frac{f(\rho)}{R^4}$$

$$f(\rho) = \frac{1-\rho}{\rho}$$

$$Q = \begin{cases} 440 + 5200\phi_{ZrO_2} & 0.05 \leq \phi_{ZrO_2} \leq 0.95 \\ 615 + 1700(1 - \phi_{ZrO_2}), & \phi_{ZrO_2} > 0.95 \end{cases} \quad kJ/mol$$

*J. Wang, R. Raj, Activation energy for the sintering of two-phase alumina/ zirconia ceramics, J. Am. Ceram. Soc., 74 (1991) 1959

Formulations to model sintering of composite and FGM (cont'd)

Equivalent particle size*:
$$R = \frac{R_l}{\chi(c_{sl}, \phi_s)}$$

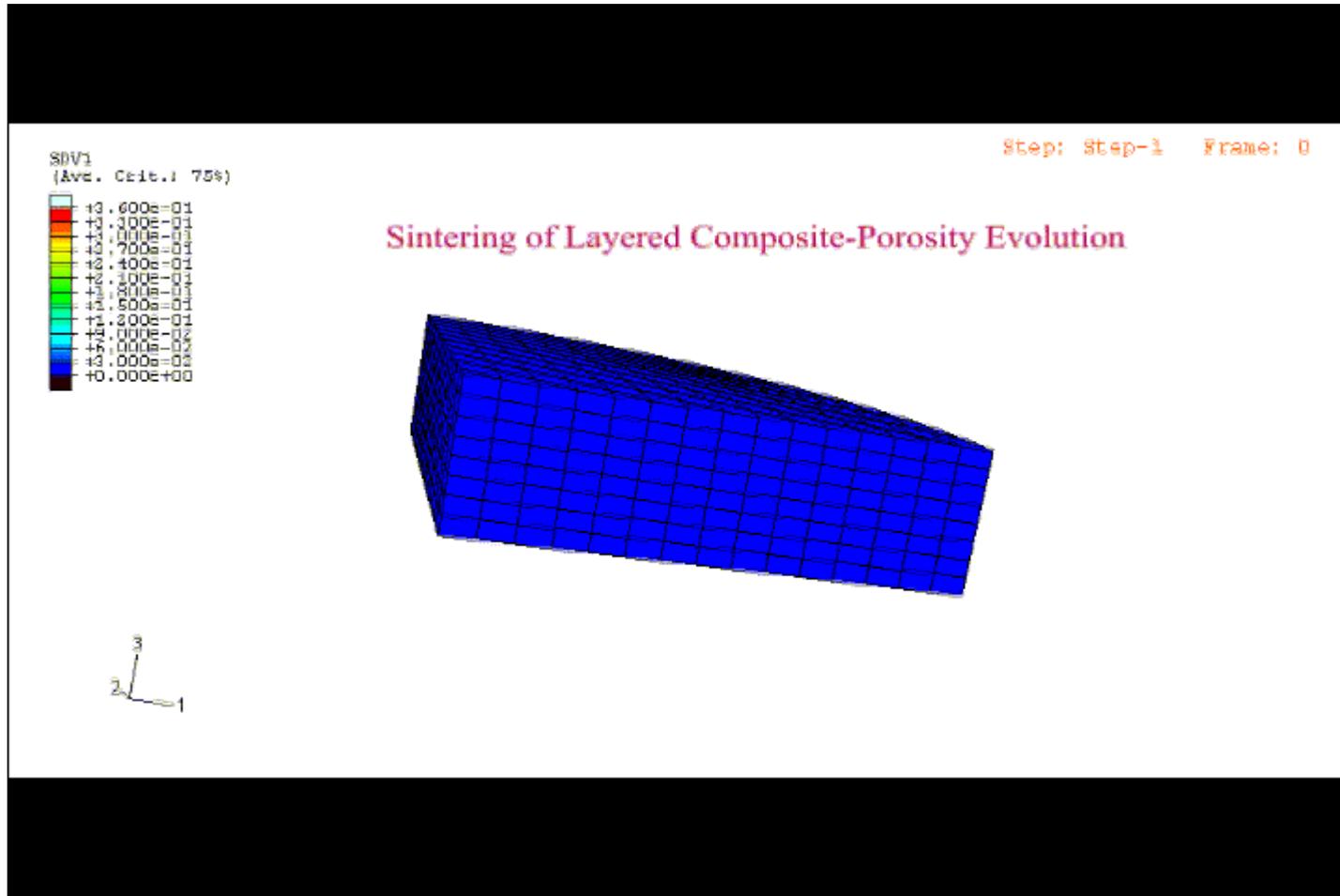
$$\chi = \frac{c_{sl}^3(1-\phi_s)^2 + \phi_s(1-\phi_s)(1+c_{sl})c_{sl} + \phi_s^3}{c_{sl}^3(1-\phi_s)^2 + 0.5\phi_s(1-\phi_s)(1+c_{sl})^2c_{sl} + \phi_s^2c_{sl}}$$

Bulk viscosity:
$$K_v = -\frac{P_L\rho}{\dot{\rho}}$$

Shear viscosity:
$$S = 1.5f(\rho)K_v = \frac{3(1-\rho)}{2\rho}$$

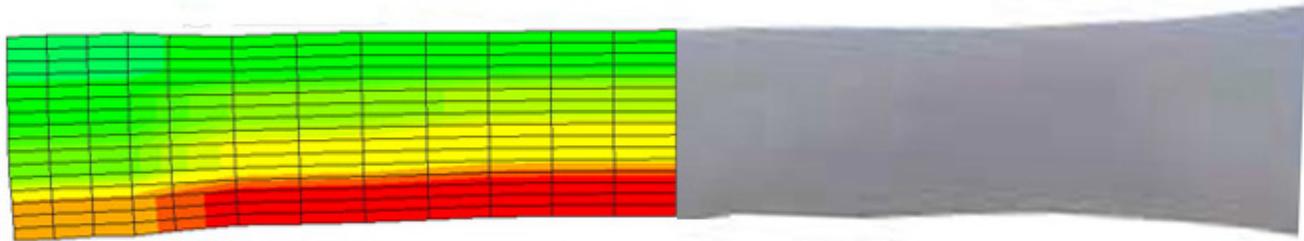
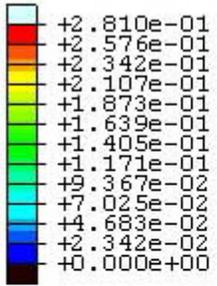
*J. Pan, H. Le, S. Kucherenko, J. A. Yeomans, A model for the sintering of spherical particles of different sizes by solid state diffusion, Acta Mater., 46 (1998) 4671.

Finite Element Modeling of shape distortion of FGM sintering

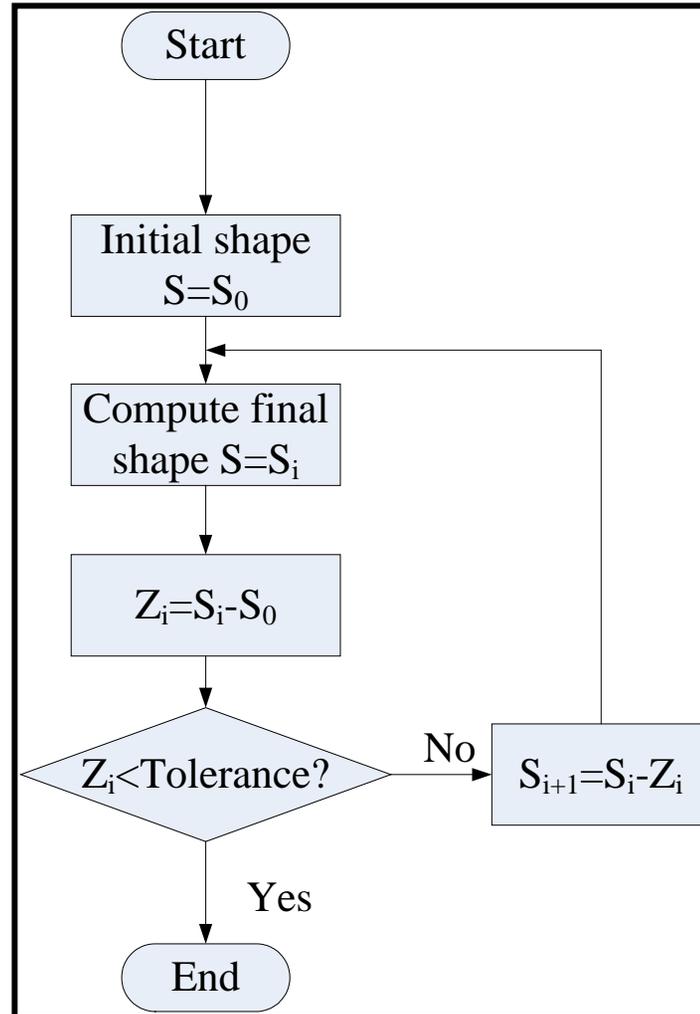


Comparison with experimental results

SDV1
(Ave. Crit.: 75%)



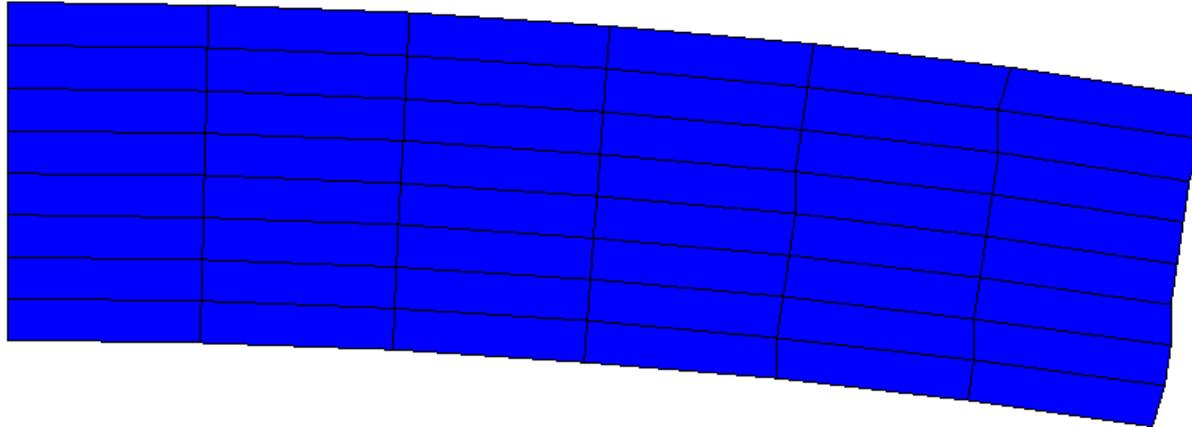
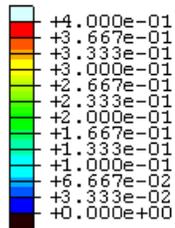
Iteration process to optimize initial shape of FGM



Inverse optimization of initial shape

Step: Step-1 Frame: 0

SDV1
(Ave. Crit.: 75%)



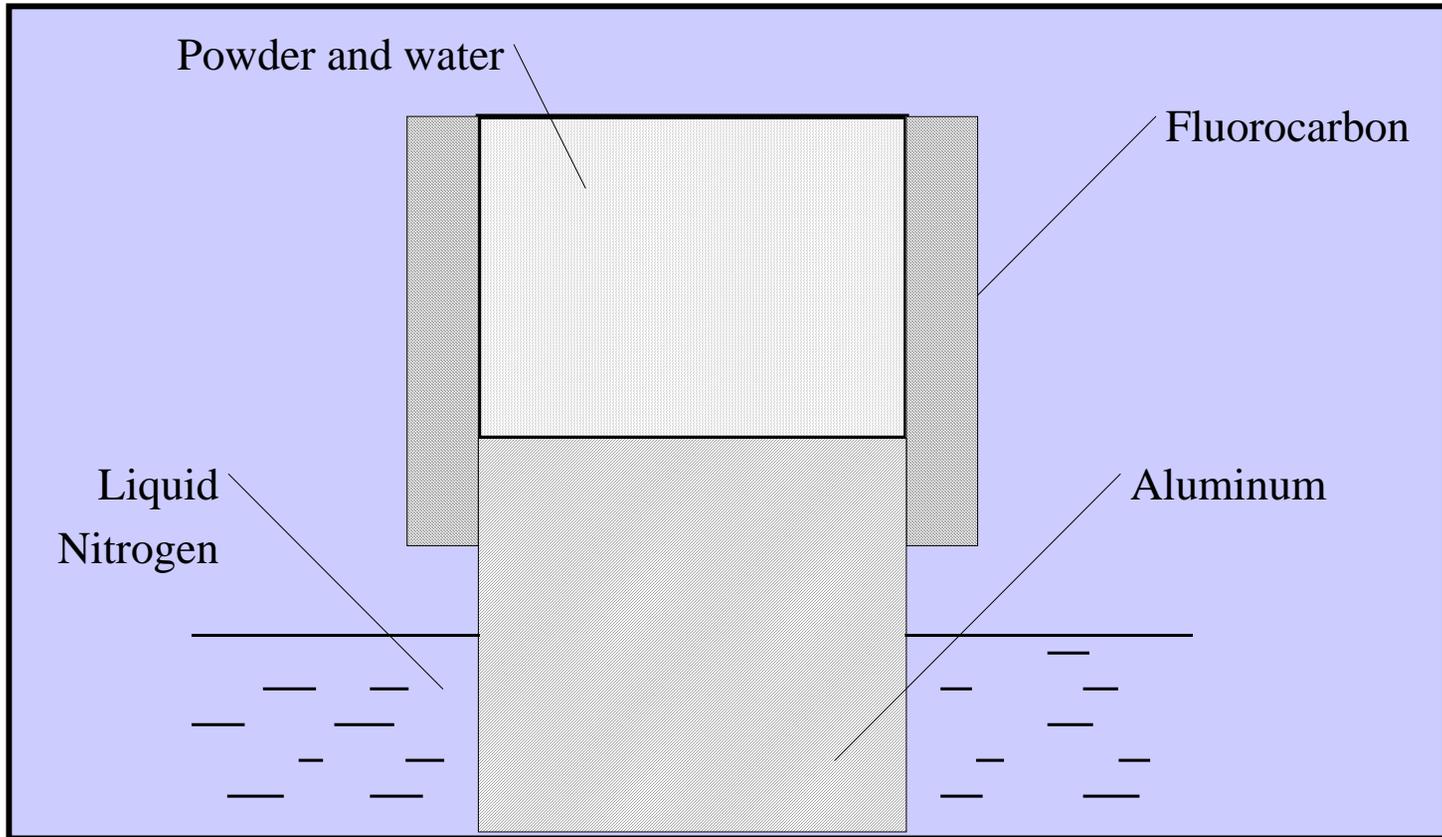
Conclusions

- **The characteristics of EPD suspension were studied. 8% vol. n-butylamine was added into the suspension to enhance particle charging. Particle agglomeration problems were solved by the ultrasonic vibration.**
- **The kinetics of the deposition of both thin films and 3-D shape components were studied. The obtained kinetics shows good agreement with Hamaker's law.**
- **The green $\text{Al}_2\text{O}_3/\text{ZrO}_2$ 3-D FGM was successfully synthesized by EPD. Disks and cylinders were deposited using a self-designed device. It was found that large particles help avoiding cracking problems during drying. The fabricated specimens were sintered and the resultant SEM micrographs show the desired graded structures.**

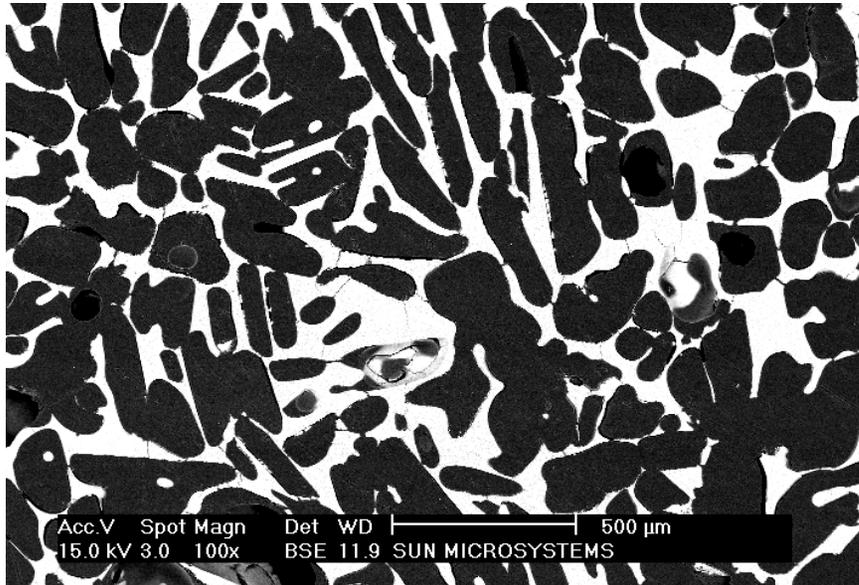
Conclusions (cont'd)

- **A user subroutine, which implements the developed constitutive formulations, was developed and linked to the commercial finite-element software ABAQUS.**
- **The sintering of a disk-shape FGM made of $\text{Al}_2\text{O}_3/\text{ZrO}_2$ was simulated. The results showed that the FGM disk has undergone warping because of the difference between the sintering kinetics of Al_2O_3 and ZrO_2 .**
- **The “inverse” methodology was successfully employed to obtain the initial shape in order to get the desired final shape after sintering.**
- **A sequential deposition process which consists of electrophoretic deposition and following electroplating was investigated. A copper sulfate plating bath was used for electroplating. An $\text{Al}_2\text{O}_3/\text{Cu}$ composite was successfully fabricated.**

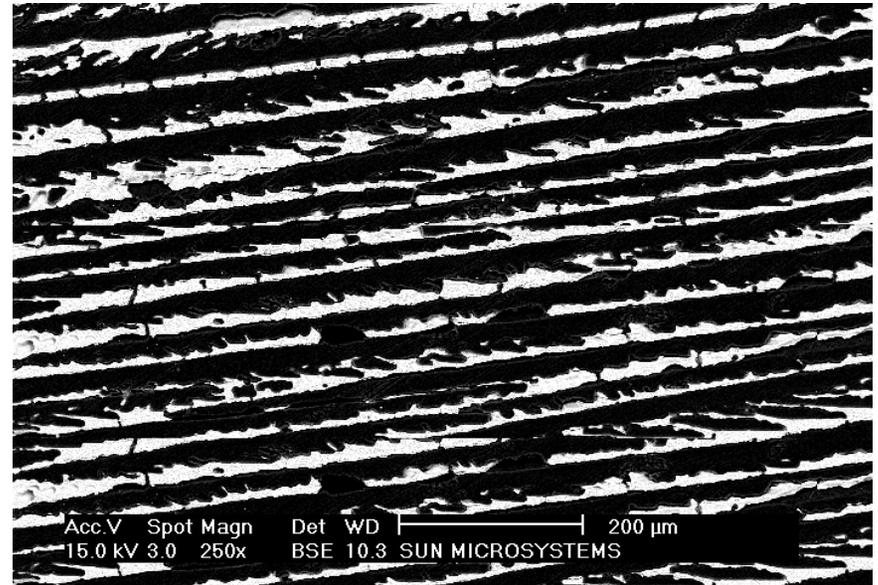
Possible future work: freeze drying



Microstructures



Regular freezing



Unidirectional freezing

CTE range of composite predicted by modeling

